

SPECIES FLAVONOIDS OF THE GENUS *Thermopsis*  
OF USSR FLORA

S. V. Teslov and L. S. Larina

UDC 547.972.23

We have investigated *Th. lanceolata* R. Br. (lanceleaf thermopsis, I, Irkutsk oblast), *Th. turkestanica* Gandogr. (II, Issyk-Kul' depression), *Th. alterniflora* Rgl. et Schmalh. (III, Tashkent oblast), *Th. dolichocarpa* V. Nik (IV, Botanical Gardens of the All-Union Scientific-Research Institute of Medicinal Plants, Moscow), *Th. fabacea* (Pall.) D.C. (bean thermopsis, V, nursery for medicinal plants, Perm), and *Th. alpina* (Pall.) Ldb. (VI, Kungei-Alatau range) in the period of their flowering.

In methanolic extracts by two-dimensional paper chromatography in the solvent systems BAW (4:1:5) (1) and 15% acetic acid (2) in I we found not less than nine, in II and IV six, in III 13, in V eight, and in VI 12 flavonoids. By chromatography on polyamide, two substances of flavonoid nature (A and B) were isolated from each of III, IV, V, and VI [1].

Substance A,  $C_{21}H_{20}O_{11}$ , mol. wt. 488, light yellow acicular crystals with mp 258-259°C (from 70% ethanol),  $[\alpha]_D^{20} -98^\circ$  (c 1.02; methanol-pyridine, 1:1),  $R_f$  0.43 (system 1) and 0.17 (system 2). A Bryant test [2] showed that it was a glycoside. With ferric chloride it gave a green coloration (free hydroxy group at  $C_5$ ).

UV spectrum:  $\lambda_{max}$  in  $C_2H_5OH$  350, 264 nm. Bathochromic shifts with the appropriate reagents showed the presence of free hydroxy groups at  $C_5$ ,  $C_3'$ , and  $C_4'$  and the absence of a free hydroxy group at  $C_7$  [3].

The hydrolysis of substance A with 10% sulfuric acid gave the aglycone  $C_{15}H_{10}O_6$  with mol. wt. 286, mp 329-330°C (from 50% ethanol); UV spectrum:  $\lambda_{max}$  in  $C_2H_5OH$  350, 267 nm; acetate with mp 226-227°C (from 80% ethanol),  $R_f$  0.85 (system 1) and 0.05 (system 2), giving a green coloration with a solution of ferric chloride. Phloroglucinol and protocatechuic acid were found in the products of alkaline degradation. The results of an analysis of the UV spectrum showed that the aglycone had hydroxy groups at  $C_5$ ,  $C_7$ ,  $C_3'$ , and  $C_4'$  [4]. The IR spectrum of the aglycone coincided completely with that of a sample of luteolin. A mixture of the latter with the aglycone gave no depression of the melting point.

D-Glucose was found in the hydrolyzate after neutralization with barium carbonate.

The above facts enabled substance A to be identified as luteolin 7-O-D-glucoside (cynaroside), which was confirmed by comparing the IR spectra of substance A and of a sample of cynaroside, and also by the absence of a depression of the melting point of a mixture.

Substance B,  $C_{15}H_{10}O_6$ , mol. wt. 286, mp 329-330°C (from 50% ethanol); acetate with mp 226-227°C (from 80% ethanol),  $R_f$  0.83 (system 1) and 0.04 (system 2). On the basis of a mixed melting point and IR and UV spectroscopy it was identified as luteolin.

Cynaroside and luteolin were detected by paper chromatography in plants I and II, as well.

From plant V a white substance was isolated with the composition  $C_{16}H_{12}O_4$ , mol. wt. 268, mp 260-261°C, acetate with mp 169-170°C. The substance was identified as pratol [5], which has been isolated from this plant previously.

---

Perm Pharmaceutical Institute. Translated from *Khimiya Prirodnikh Soedinenii*, No. 5, pp. 666-667, September-October, 1973. Original article submitted January 25, 1973.

© 1975 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

#### LITERATURE CITED

1. V. A. Bandyukova, *Rast. Res.*, 596 (1965).
2. E. F. Bryant, *J. Amer. Pharmac. Assoc.*, 480 (1950).
3. V. I. Litvinenko and N. P. Maksyutina, *Khim. Prirodn. Soedin.*, 420 (1965).
4. N. P. Maksyutina and V. I. Litvinenko, *Phenolic Compounds and Their Biological Functions* [in Russian], Moscow (1968), p. 7.
5. G. Klein, *Handbuch der Pflanzenanalyse III/4 Sp. A. II*, Wien, (1932), p. 862.
6. A. A. Ryabinin and E. M. Il'ina, *Zh. Prikl. Khim.*, 28, 663 (1955).